

## Study Guide for CHEM3420 Test #1

Fall

Material covered: You are responsible for anything we have discussed in class as well as anything that is in the book Ch. 8 (all) and Ch. 9 through p. 285 (stop right before “d: Degeneracy”). This study sheet is a guideline to help you sort through the numerous topics that you have covered. You are also responsible for anything that has been on your assignments and your homework.

Information that you will have at your disposal:

- Periodic table, scratch paper, various integral solutions
- You may also bring a 4X6” with constants and formulas only written on both sides.

The make-up of the test: multi-match, short-answer, discussion, short and long problems

### I. The history of Quantum Mechanics

#### A. The people of QM

**You need to know who they were, what they did (in word and formulae), and how what they did contributed to QM.**

Balmer  
Bohr  
Schrodinger  
Einstein  
Pauli  
Heisenberg  
Dulong  
Petit  
Dirac  
Planck  
Wien  
deBroglie  
Rydberg  
Angstrom  
Rayleigh

#### B. The events leading up to/phenomena involved in the discovery of QM

### II. Math

#### A. Concepts that we have used

**Be able to use these and manipulate them.**

Complex numbers, complex conjugate  
Rules of exponents  
Eigenfunctions, eigenvalues  
Operators  
Derivatives  
Integrals  
Partial derivatives  
Triple integrals  
Coordinate systems

#### B. Specific things you need to know

How to take the derivative of any function  
How to do simple integrals (e.g.  $\int x dx$ ,  $\int \sin(x) dx$ ,  $\int e^x dx$ ), etc.  
How to look up integrals in an integral table  
Any math involved in any homework or quiz problem.

#### IV. QM concepts

These are general topics. Many of them include definitions, formulas, problems, etc. You need to know it all!

Schrodinger's equation:

1-D PIB: the Hamiltonian, the wavefunction, the energy

3-D PIB: the Hamiltonian, the wavefunction, the energy

Be able to find the energy by operating on the wavefunction with the Hamiltonian given a wavefunction.

~~Heisenberg uncertainty principle~~

Normalization of a function

Expectation values

Orthogonalization of a function

Wave-particle duality

The postulates of QM

Orthonormal complete set

Requirements for a good wavefunction

#### V. Suggested problems at the ends of the chapters to work.

Chapter 8

Exercises: 1b, 2b, 4b, 7b, ~~12b~~

Problems: 5 'a'. (Change 'sphere' to 'confined to move on a circle'), 8, ~~13 a, b, c (not d)~~, 14, 20

Chapter 9

Exercises: 3b